

Thermochemische Verfahren zur energetischen und stofflichen Nutzung von Biomasse

Nicolaos Boukis, Nicolaus Dahmen, Axel Funke, Andrea Kruse, Klaus Raffelt, Jörg Sauer, Hamm, 22.07.2015

Institute of Catalysis Research and Technology - Project bioliq



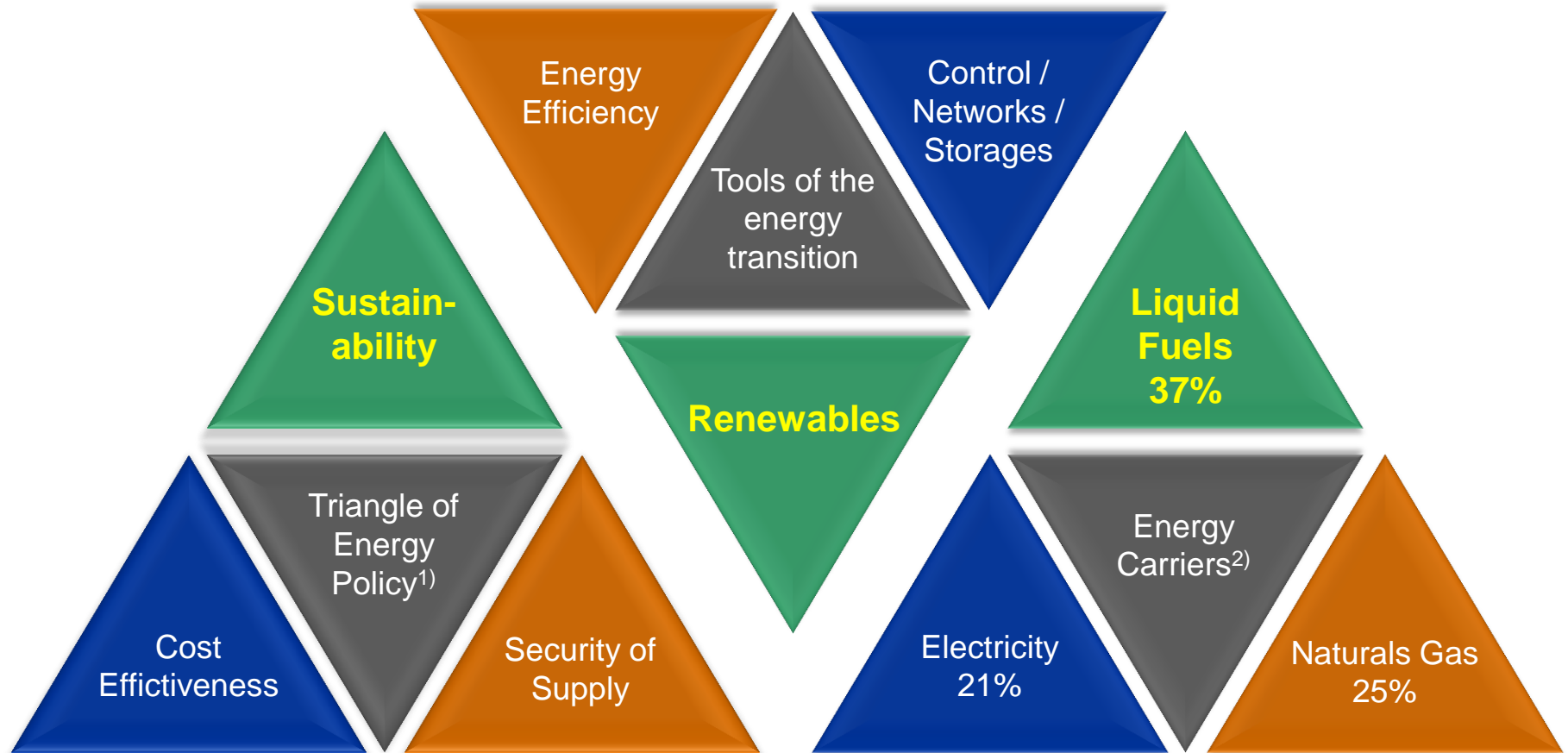
Outline

- Motivation
- Hydrothermal Processes, Hydrothermal Gasification
- The bioliq Approach to BtL
- Conclusion and Outlook

Outline

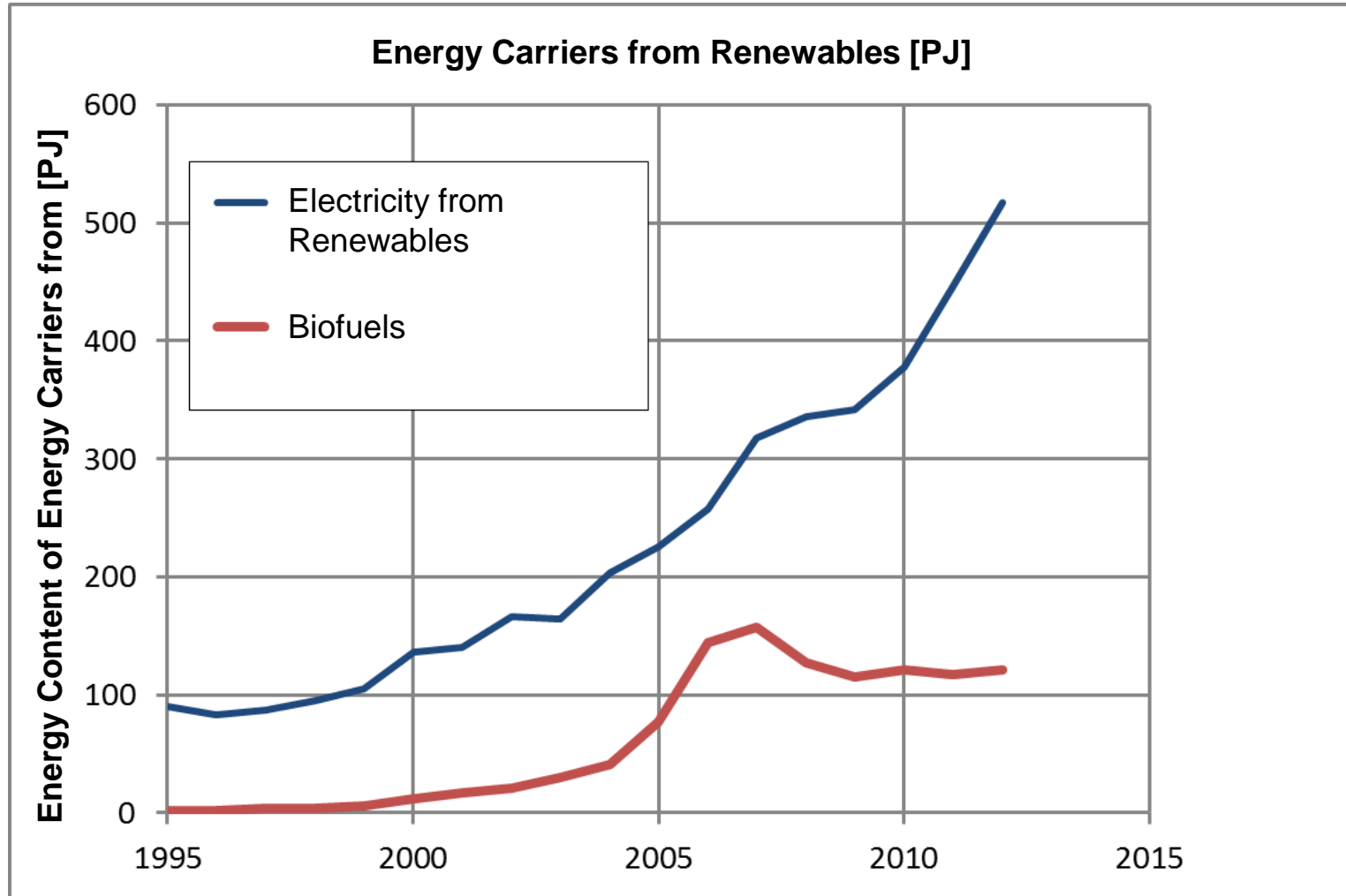
- Motivation
- Hydrothermal Processes: Hydrothermal Gasification
- The bioliq Approach to BtL
- Conclusion and Outlook

„Magic Triangles“ of the German Energiewende



- 1) Zweiter Monitoring-Bericht „Energie der Zukunft“, Bundesministerium für Wirtschaft und Energie (BMWi), Berlin, März 2014
- 2) AGEb, Arbeitskreis Energiebilanzen, September 2014, www.ag-energiebilanzen.de/ entnommen am 21.09.2014

„Energiewende in Germany“ – Biofuels and Electricity¹⁾



1) AGEb, Arbeitskreis Energiebilanzen, September 2014, www.ag-energiebilanzen.de/ retrieved on 2014-09-21

Erdöl - Zahlen

- Erdöl Tagesproduktion: 83 Mio bbl per day
= 13,2 Mio m³/Tag
- Zum Vergleich:
Speichervolumen des Brombachsee 164 Mio m³
 - Zeit bis zum Auffüllen des Brombachsees mit weltweiter Erdölproduktion: 12 Tage 10 Stunden



Source: IEA, Energy Technology Perspectives 2014
<http://www.landeskraftwerke.de/brombachsee.htm>, entnommen am 14.07. 2014
<http://www.zv-brombachsee.de/>, entnommen am 14.07. 2014

Noch mehr Zahlen

- | | |
|-----------------------------------|---------------------------|
| ■ Investitionssumme für Pearl GTL | 18,5 Mrd USD |
| Produktionsmenge | 0,14 Mio bbl per day |
| Anteil an der Welt-Ölproduktion | 0,17% |
| Investkosten pro bbl | 130.000 USD/(bbl per day) |
| Mitarbeiter für Anlagenbau: | ca. 50.000 |

- | | |
|---------------------------------------|------------------------------|
| ■ Hochrechnung auf Weltöl-Produktion: | |
| Kosten: | 10,8 Billionen USD |
| Investbudget der Ölkonzerne (2012): | 260 Mrd USD (x 41) |
| ExxonMobile (2013): | 42,4 Mrd USD (x 254) |

- | | |
|----------------------------------|------------------------|
| ■ Hochgerechneter Personalbedarf | ca. 30 Mio für 5 Jahre |
|----------------------------------|------------------------|

Source: <http://www.platts.com/>
<http://www.shell.com/>
<http://ir.exxonmobil.com/>
<http://energypolicy.columbia.edu/>
 entnommen am 13.07.2014

Contribution of the Negative Residual Load

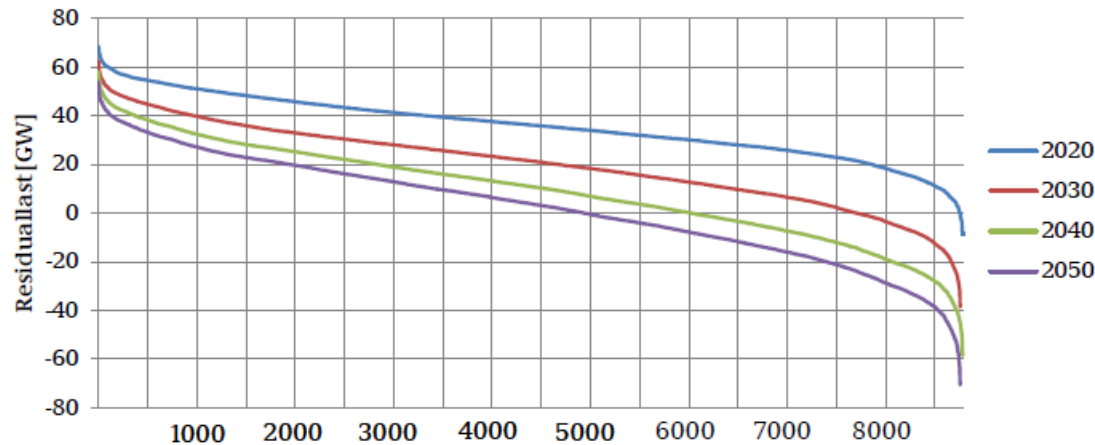


Abbildung 4-34: Geordnete Dauerlinie der Residuallast.

Negative residual load:	66TWh = 240 PJ
After conversion losses:	33TWh = 120 PJ
Liquids consumption Germany 2012:	1290TWh = 4640 PJ

Potential for Liquids from residual load Availability	2,2 Mio to (2,6%) max. 3000h
---	---------------------------------

Sources: Deutsche Energie-Agentur GmbH (dena), Integration der erneuerbaren Energien in den deutschen/europäischen Strommarkt, 2012
AGEB, Arbeitskreis Energiebilanzen, September 2014,
www.ag-energiebilanzen.de/ entnommen am 21.09.2014

Feedstocks for Future Bionenergy Applications

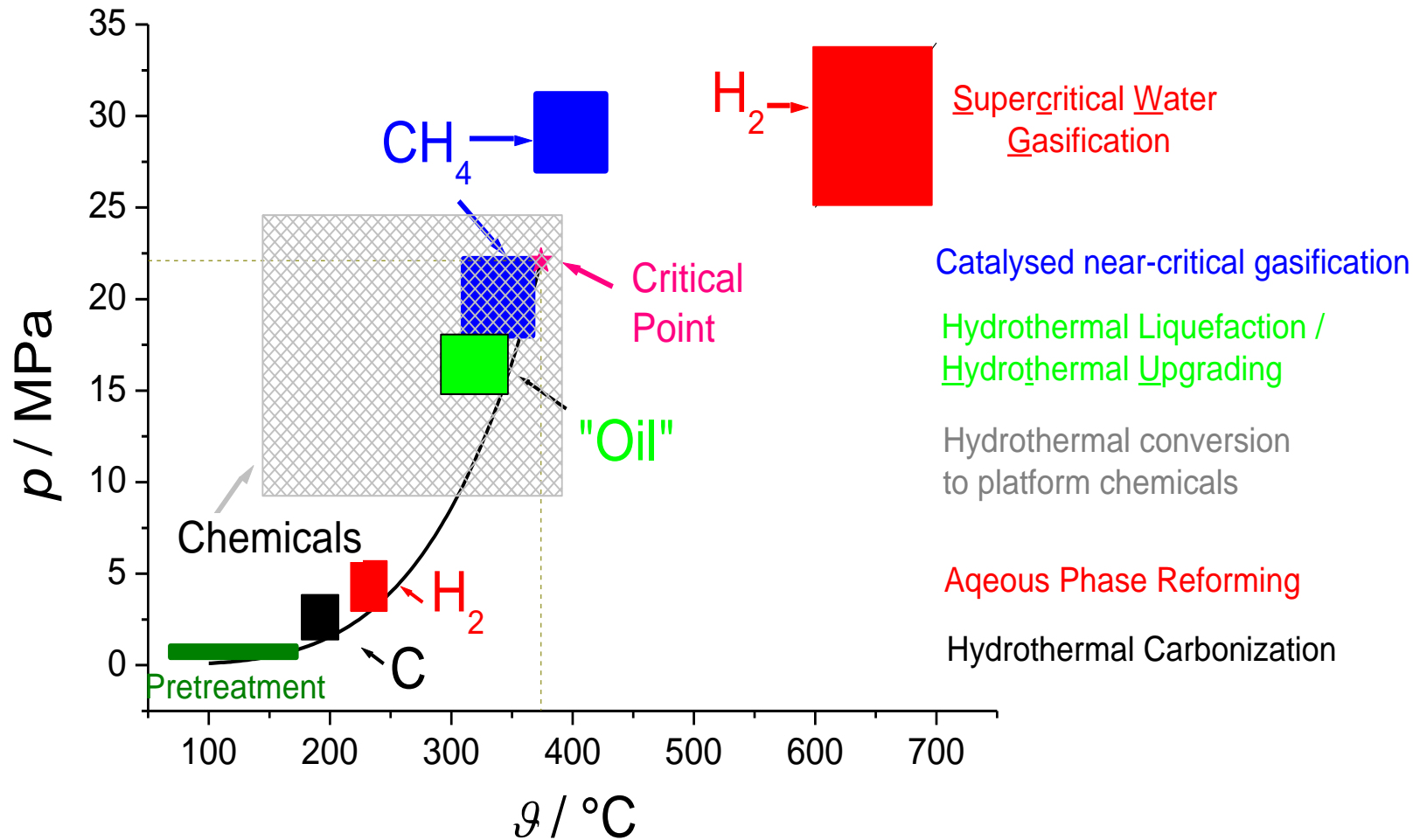
- Agriculture
 - Straw, hay,
 - Energy crops
- Forestry
 - Residues (brush, tops, stumps)
 - Thinnings
 - Short rotation plantation
- Marginal Farmland
 - Streets, railway tracks
 - Power transmission lines
- Organic residues
 - Recovered waste wood
 - Organic waste fractions
- Algae



Outline

- Motivation
- Hydrothermal Processes: Hydrothermal Gasification
- The bioliq Approach to BtL
- Conclusion and Outlook

Regimes for Hydrothermal Conversion

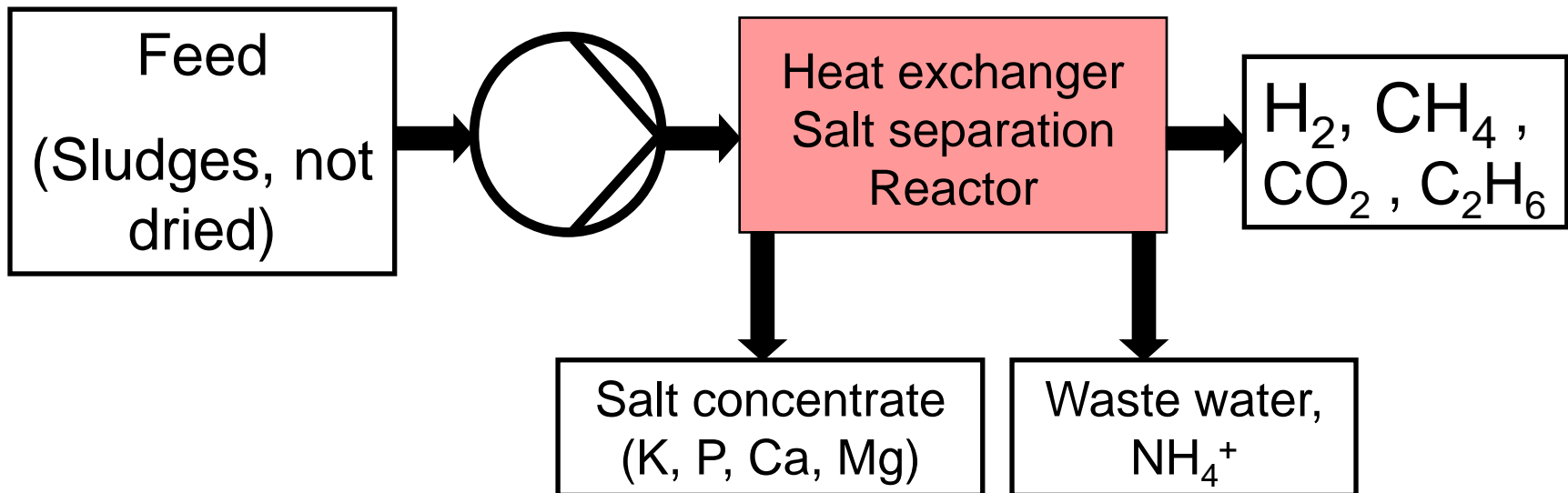


Kruse et al., 2013, modified

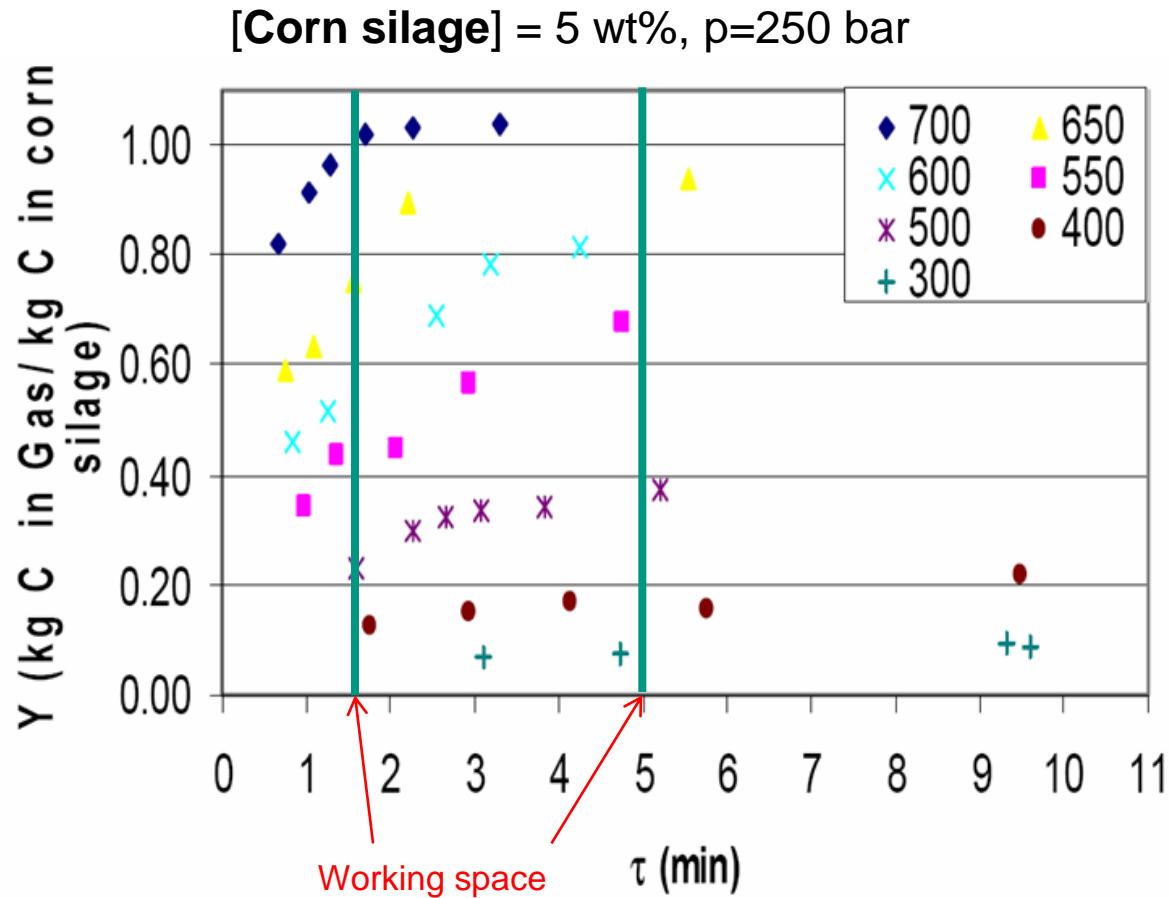
Gasification of Waste Biomass and Organic Waste Fractions

Process

Conditions: $T \approx 650 \text{ }^\circ\text{C}$; $p \approx 280 \text{ bar}$



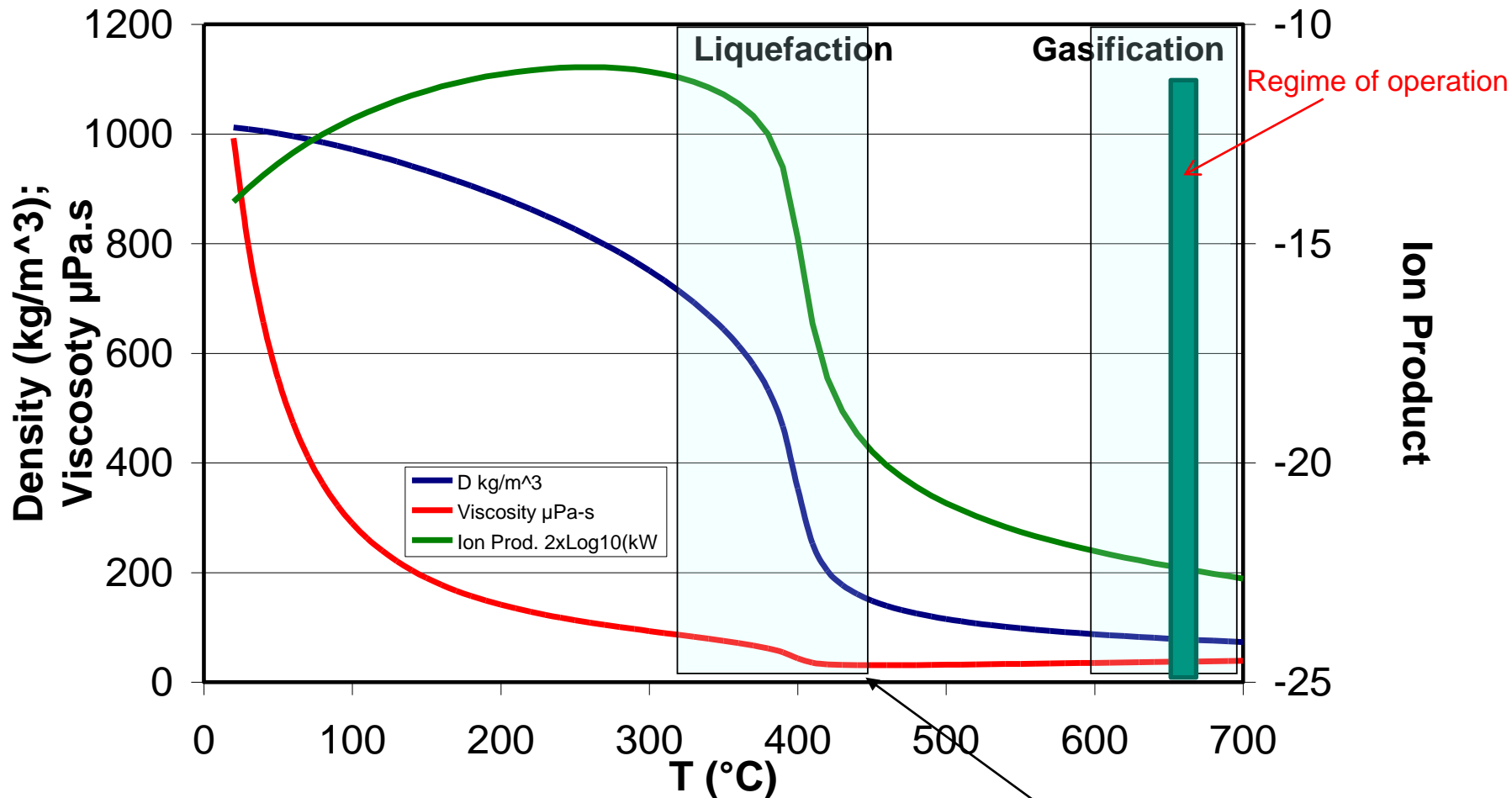
Gasification Yield as Function of the Residence Time and the Reaction Temperature



P. D'Jesus, N. Boukis, B. Kraushaar-Czarnetzki, E. Dinjus. Ind. Eng. Chem. Res. 2006, 45, 1622-1630

Properties of Water at High Temperatures and Pressure ($T_c=374\text{ }^\circ\text{C}$, $p_c=221\text{ bar}$)

Water; isobar 300 bar

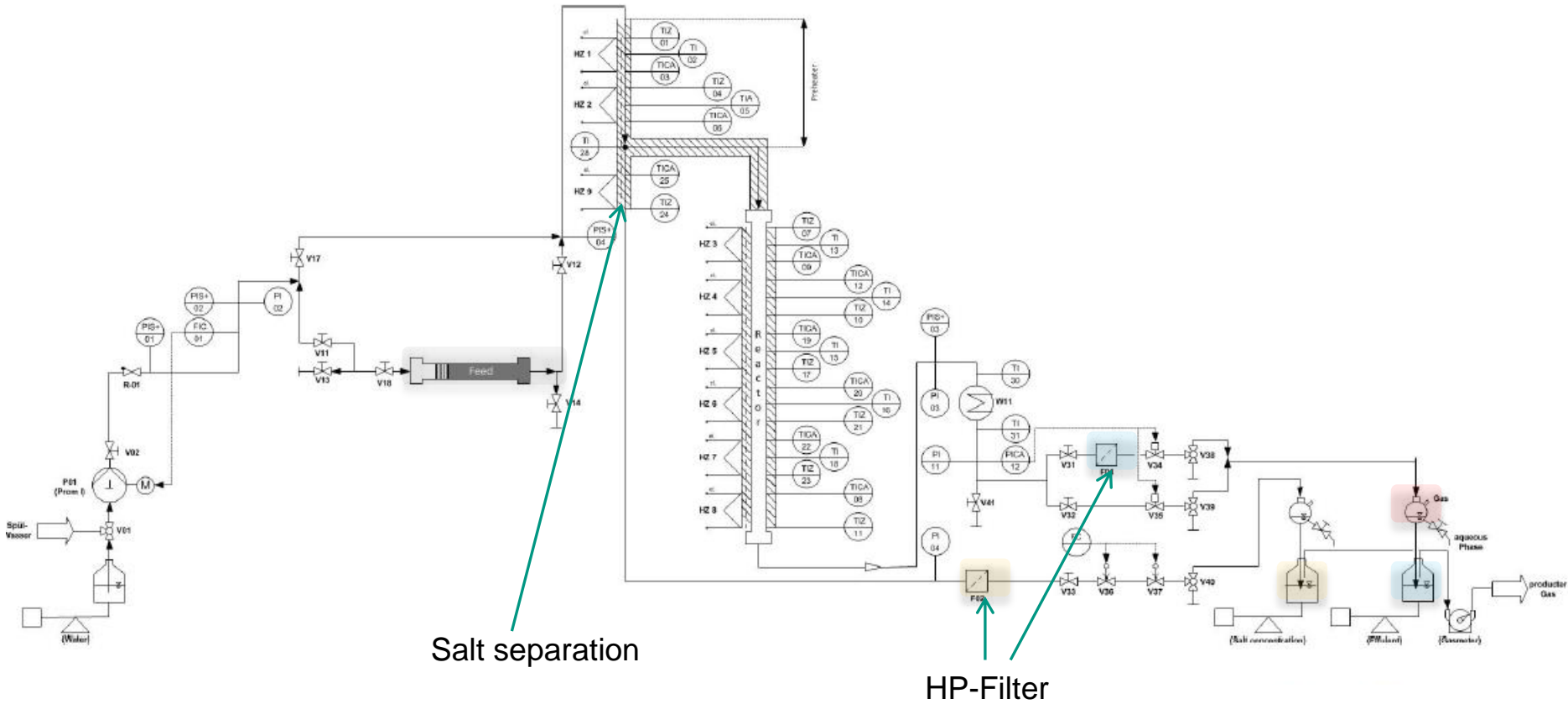


Direct heat exchange is possible => no drying

Salt precipitation

LENA – Test Rig, continuous flow,

$T_{\max} = 700\text{ }^{\circ}\text{C}$, max. flow rate 2 l/h, $V_{\text{Reactor}} = 270\text{ cm}^3$



Sewage Sludge from Oijen and Lelystad

Type	w_{DM} Dry matter, wt.%	Ash content by 550 °C, wt.%
Lelystad	17,5	18,8
Oijen	27,4	20,40

Parameter	Feed analysis	
	wt. [%]	wt. [%]
	Oijen	Lelystad
TC	41,6	43,5
TIC	0,1	0,1
TOC (C)	41,5	43,4
H	4,22	6,37
N	4,22	7,28
P	2,4	3,16
S	0,78	1,11
Ca	2,25	1,4
K	0,36	1,14
Mg	0,31	0,72
Na	0,088	0,105
Si	2,65	2,63
Al	1,22	0,56
As	0,02	0,02
Cd	0,01	0,01
Cu	0,56	0,018
Pb	0,011	0,02
Zn	0,086	0,032
Cr	0,0039	0,01
Fe	1,38	0,38
Mo	0,02	0,02
Ni	0,02	0,02

Main results, steady state operation

Total sewage sludge treated 2 – 3 kg per experiment

Type	Time steady state	Feed steady state	C _{total} steady state	Plugging	Y _{Gas}	Y _C	TOC-destruction	TOC-waste water	NH ₄ ⁺ -waste water	TN _b -waste water
[-]	[h]	[g]	[g]	[after h]	[%]	[%]	[%]	[mg/l]	[mg/l]	[mg/l]
Lelystad	5	2115	109,18	No	67,95	80,67	97,8	1953	11724	9300
Oijen	3,93	1656,5	88,86	6,5	71,21	81,83	96,7	2141	8821	5580

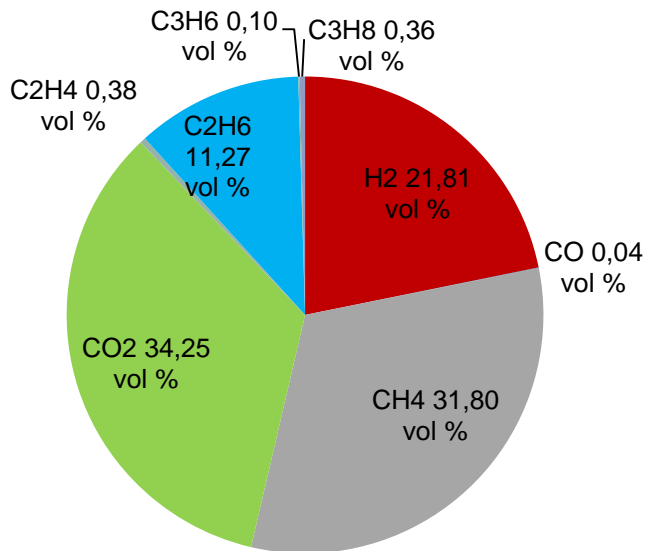
Carbon balance during steady state operation is 80 % an indication of accumulation of Carbon in the system (insoluble carbonates in the filter cakes)

Hydrothermal Gasification of Sewage Sludge

Sewage sludge	T _{Reaction}	Concentration	mean res. time
type	[°C]	[wt. % DM]	[min]
Lelystad	653	11,69	2,75
Oijen	649	12,71	2,76

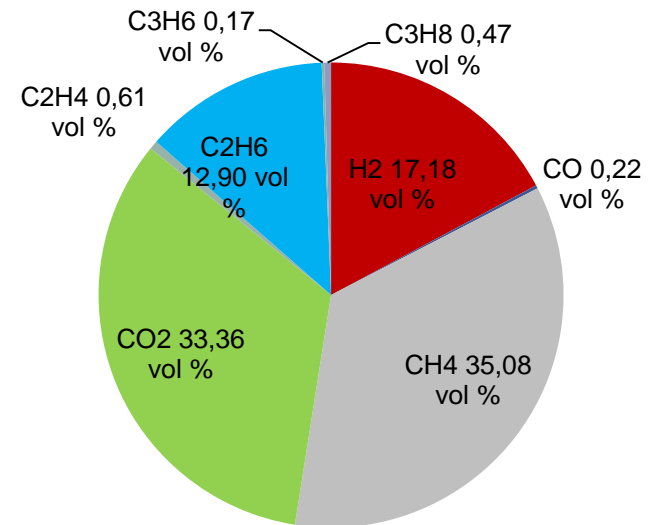
Type Lelystad

Gas composition



Type Oijen

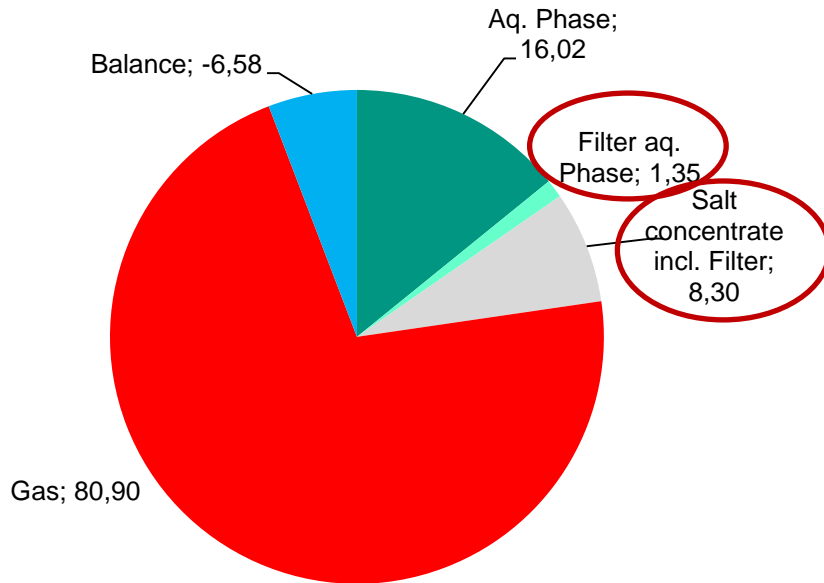
Gas composition



The Carbon Balance

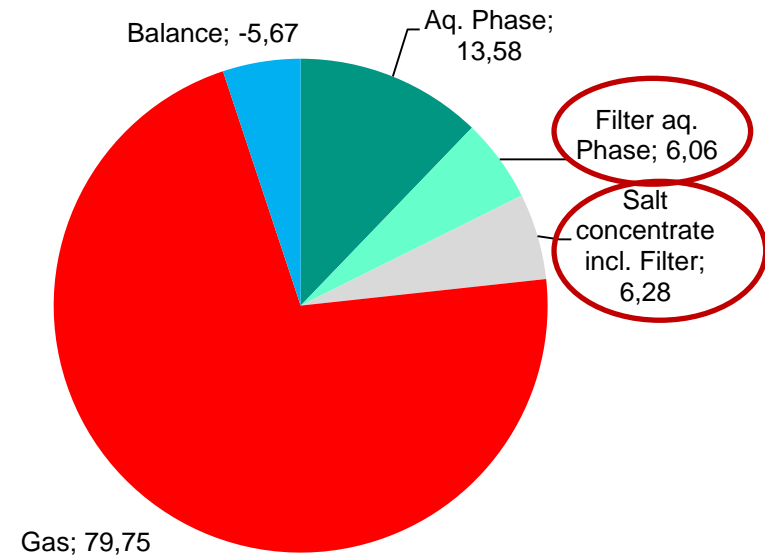
Type "Lelystad"

Carbon distribution %



Type "Oijen"

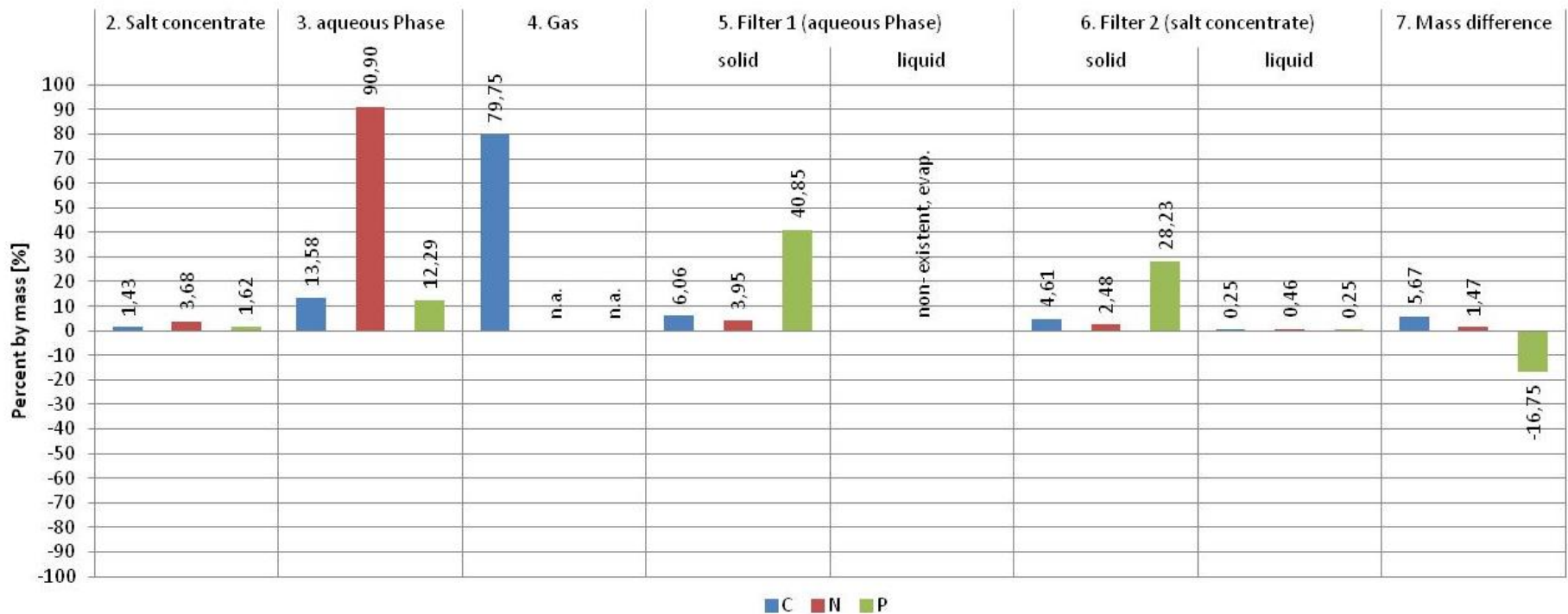
Carbon distribution %



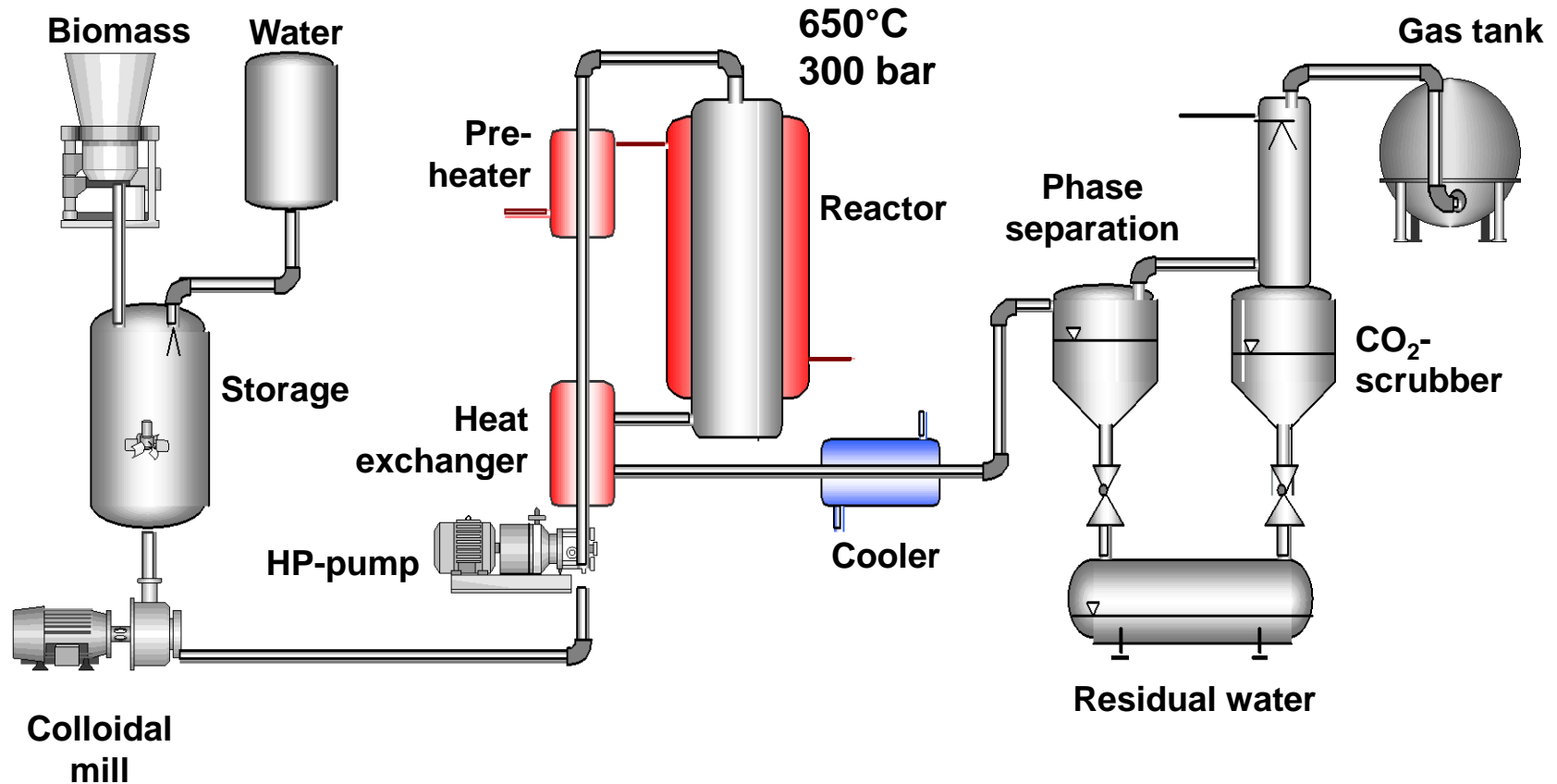
The carbon balance for the whole experiment is much better.
About 10 % of the Carbon forms carbonates

Type Oijen C, N, P Detailed Analysis

Percentage distribution of no- metal elements after gasification
(Integrated Assessment)



Scale-up to Pilot Scale (1)



VERENA pilot plant at KIT

Scale-up to Pilot Scale (2)

Feeding

100 kg / h
5-20 % dmc

Reactor

35 L volume
0,11 m i.D., 3,7 m length
Inconel Ni-Alloy
External Heating
35 Mpa; 700 °C



Conclusions for Hydrothermal Gasification

- Stable operation with “difficult feedstock” sewage sludge is possible
- Good carbon balance in lab scale ($100 \pm 10 \%$)
- High gasification yield (80 %) and η (up to 0.9). Acceptable efficiency at 12 wt.% DM possible

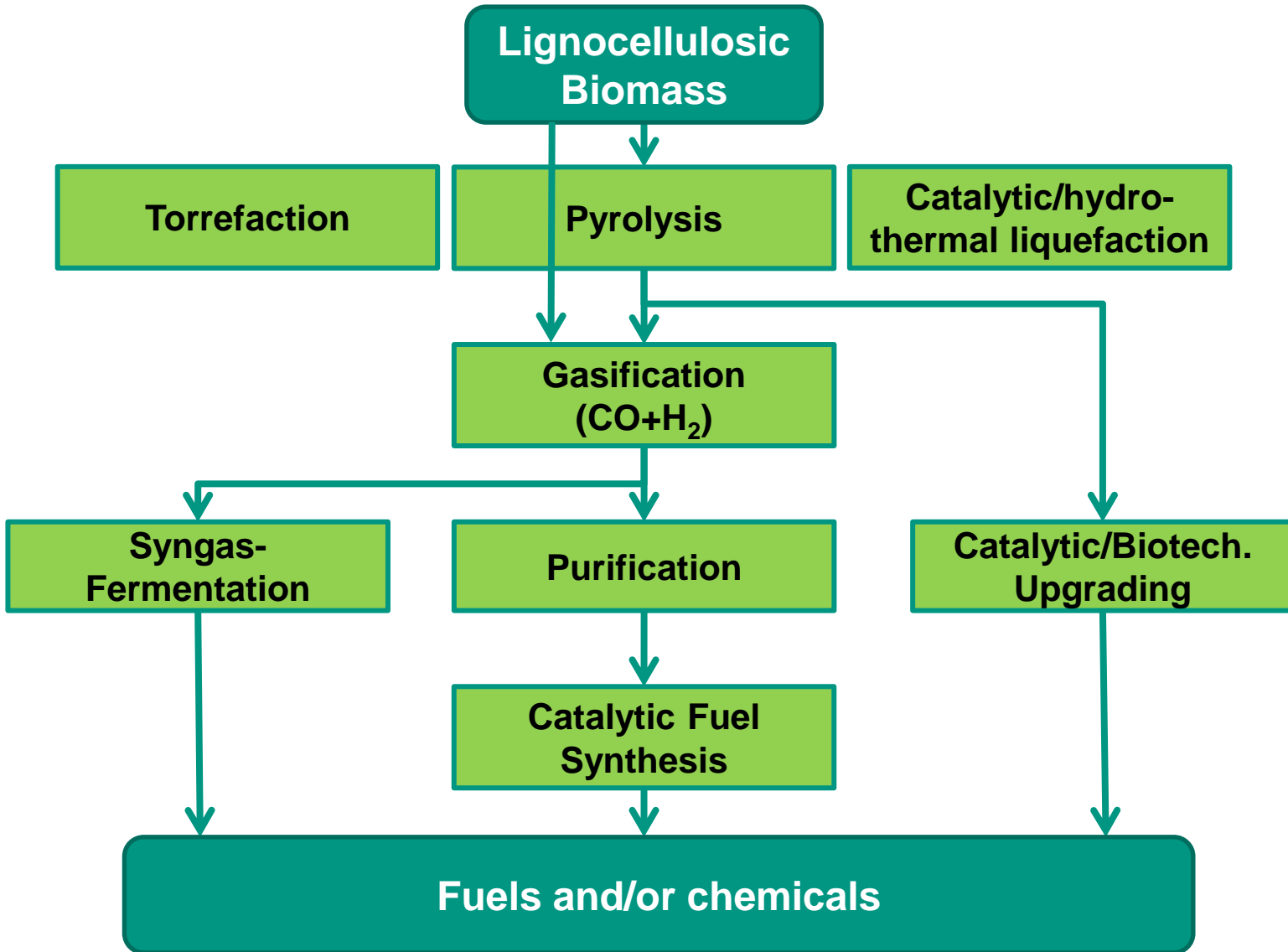
Challenges

- Salt separation is of utmost importance
 - Fouling/ scaling e.g. in heat exchangers
 - Corrosion
- HTG of sewage sludge operates at the frontier of development modern materials of construction
- Suitable models for reaction kinetics and gasification reactor are missing

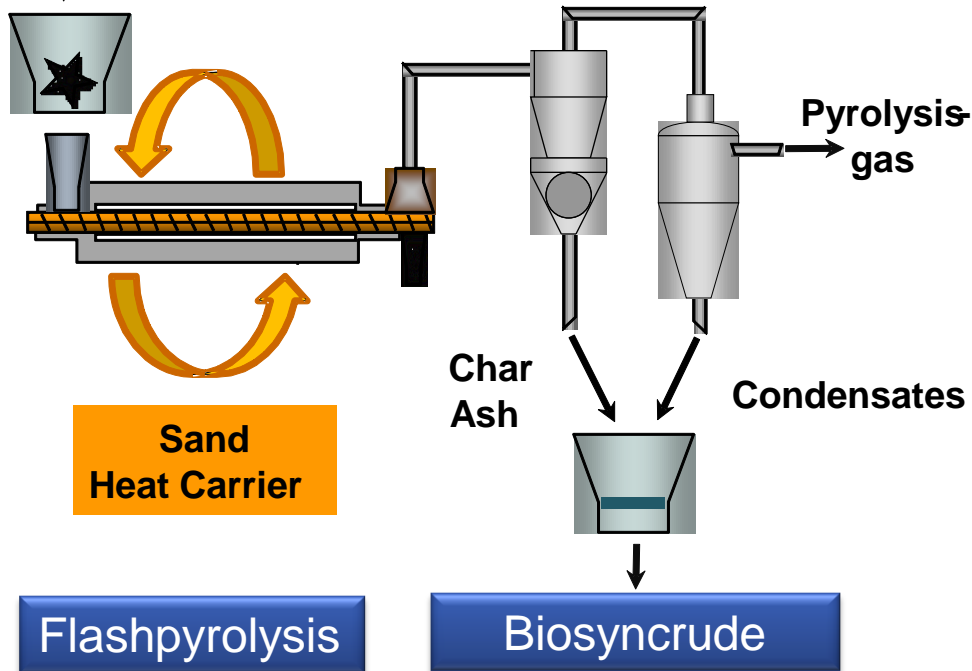
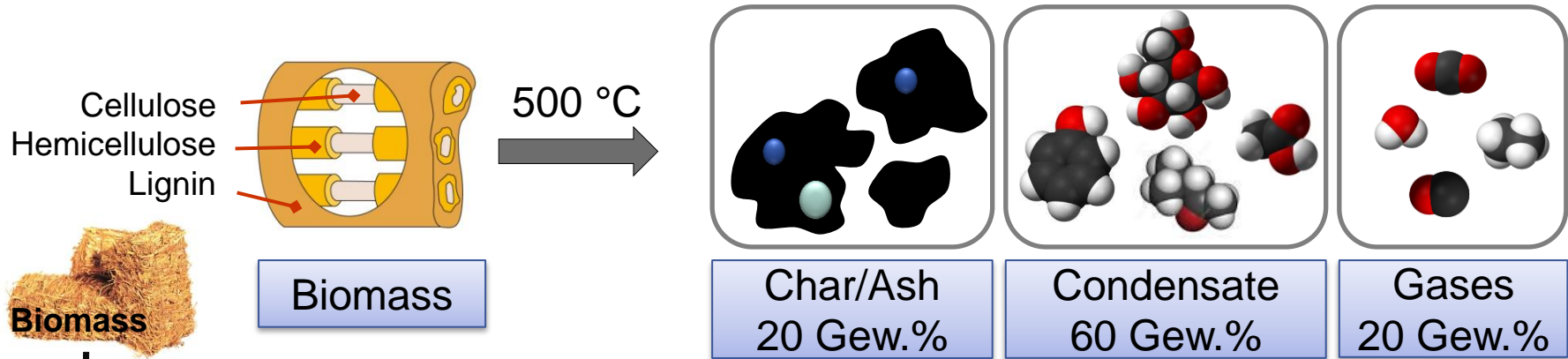
Outline

- Motivation
- Hydrothermal Processes: Hydrothermal Gasification
- The bioliq Approach to BtL
- Conclusion and Outlook

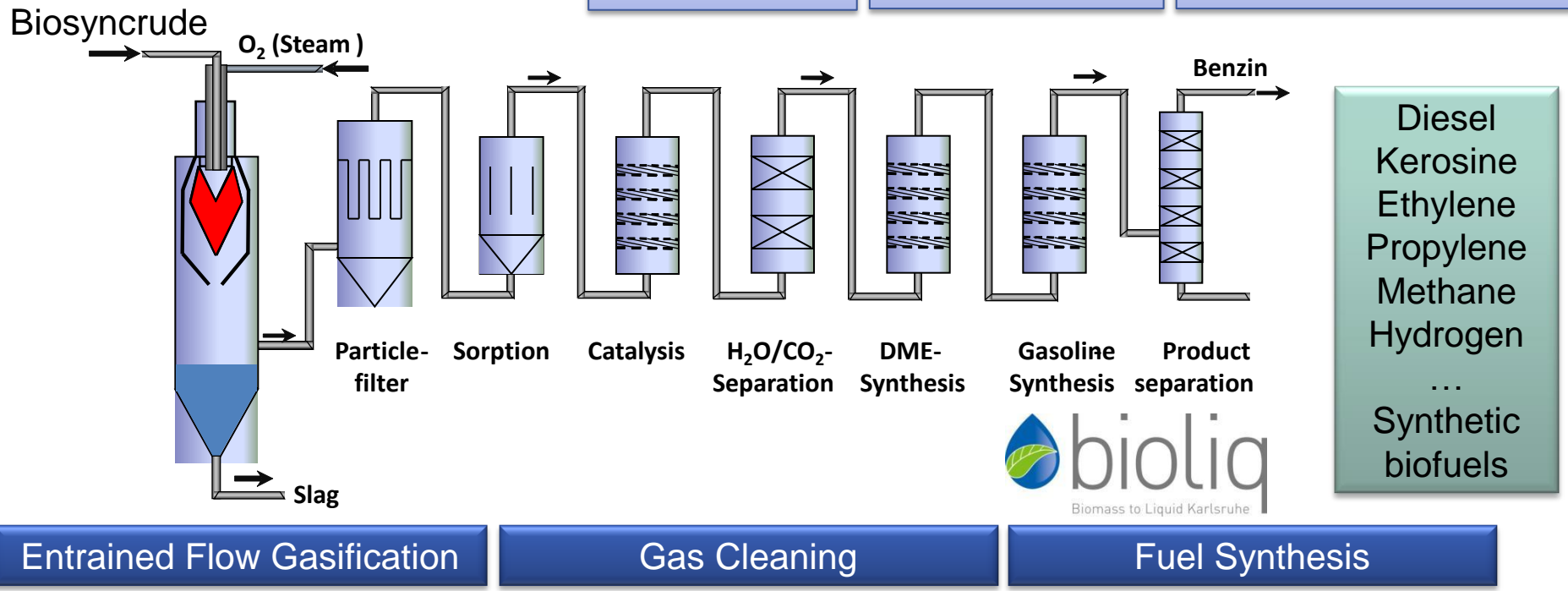
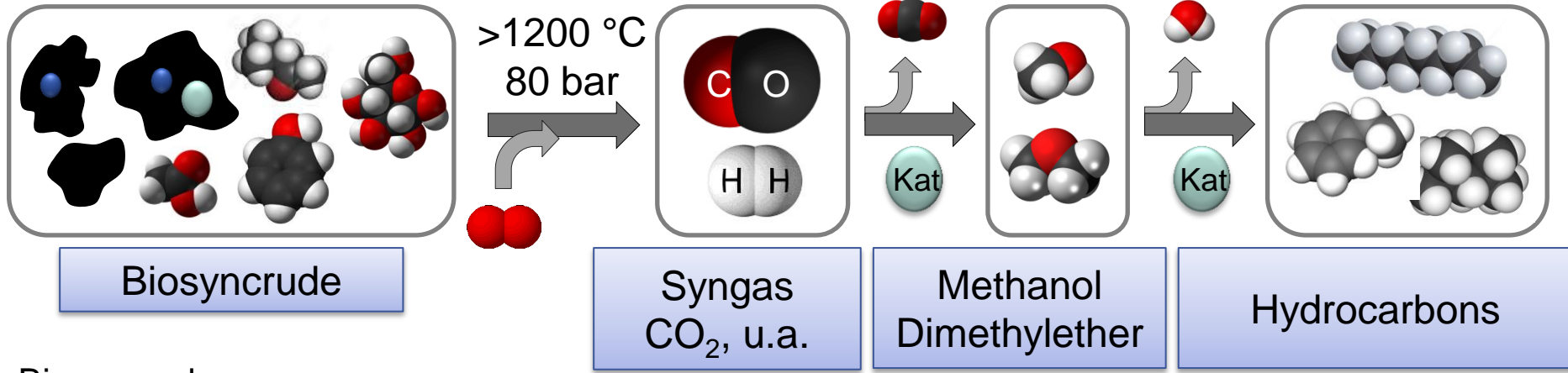
Thermochemical BtL Value Chains



Chemistry and Technology – decentralized



Chemistry and Technology – centralized



bioliq®-Pilot Plant at KIT

Fast Pyrolysis
Biosyncrude-Production

Gasification
Syngas-Production

Gas-Cleaning and
Fuel Synthesis



Technical Validation

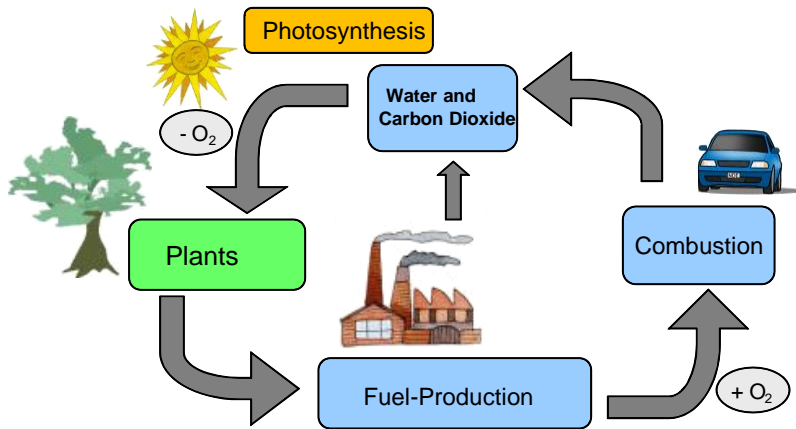
&

Platform for Research

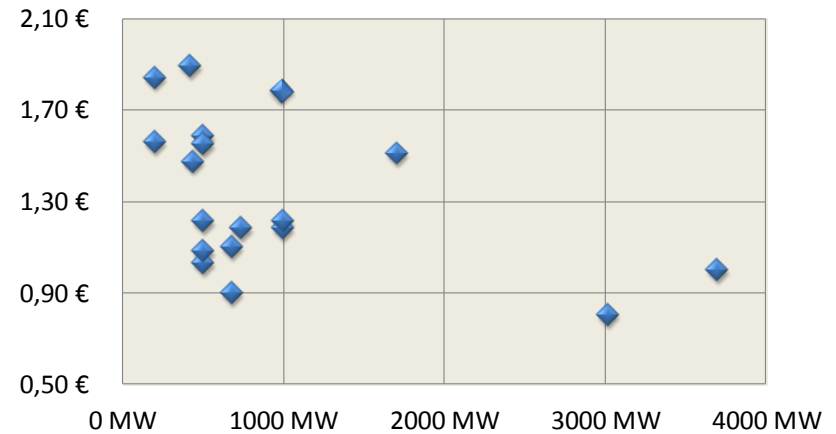
- Mass and energy balances
- Scale-up
- Stability and availability
- Production costs

- Improved insights in processes
- Optimization and development
- Diagnostics, modelling, simulation
- New applications of products

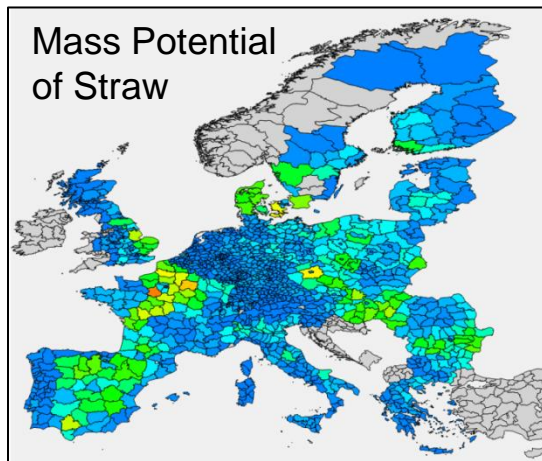
Contributions to the Analysis on the System Level



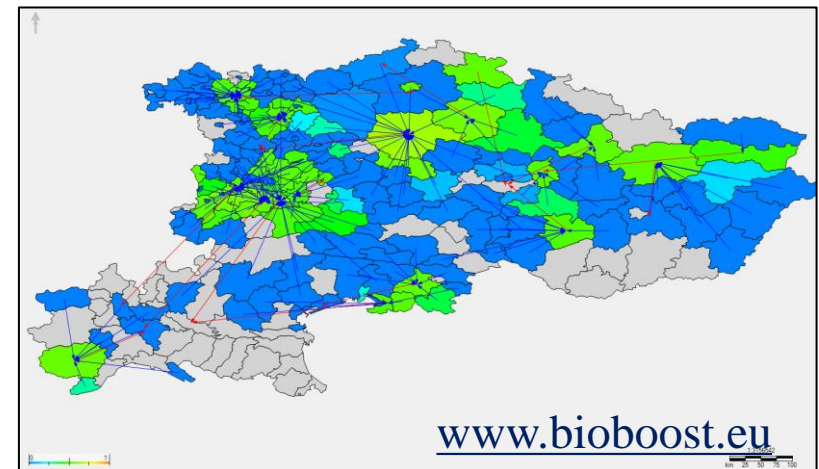
CO₂ reduction potential > 80 %



Production Costs



Potentials of sustainable supply



Logistics simulation and production network

Next steps

- Improve availability of pilot plants
- Process optimization and further development
- Development of business models & market implementation plans
- Creation of a consortium for the development of high performance fuel components



Gefördert durch:



aufgrund eines Beschlusses
des Deutschen Bundestages



Investition in die Zukunft
gefördert durch die Europäische
Union Europäischer Fonds für
regionale Entwicklung und das
Land Baden-Württemberg



Outline

- Motivation
- Hydrothermal Processes: Hydrothermal Gasification
- The bioliq Approach to BtL
- Conclusion and Outlook

The Future of Bioenergy Research at KIT (1)

„Integration into the Energy System“

bioliq®

Biomass
(eg. straw)

Flash
pyrolysis

Entrained
flow gasi-
fication

Gas
cleaning

Catalytic
synthesis

Fuels, chemical
energy carriers



Syngas-
Fermentation

Intermediates /
Biomaterials

Fischer-Tropsch
Synthesis

Fuels, chemical
energy carriers

Methanation

Gasturbine
+ Generator

Electricity

EnergyLab
2.0



Photobio-
reactor

Cultivation/
harvesting

Elektro-
poration

Product
Separation

Hydro-
thermal
liquification

Intermediates /
biomaterials

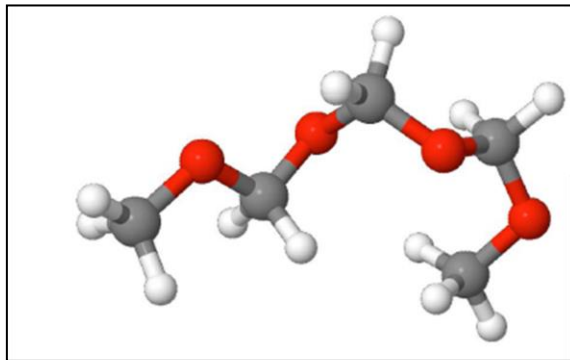
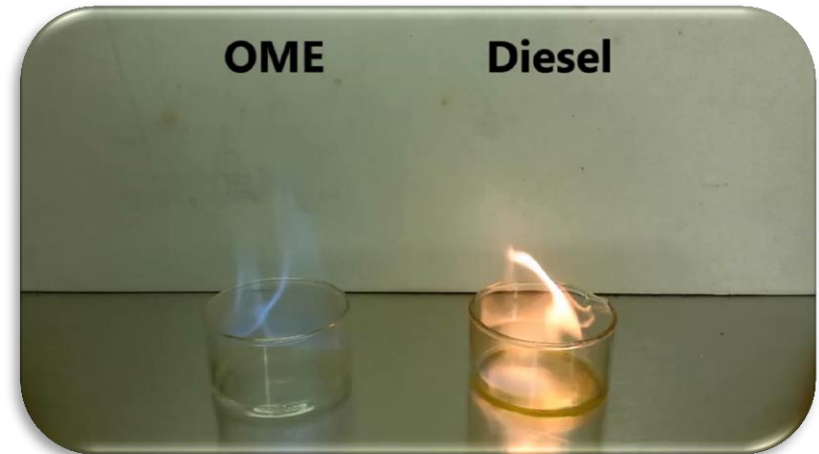
Algae Value Chain

The Future of Bioenergy Research at KIT (2)

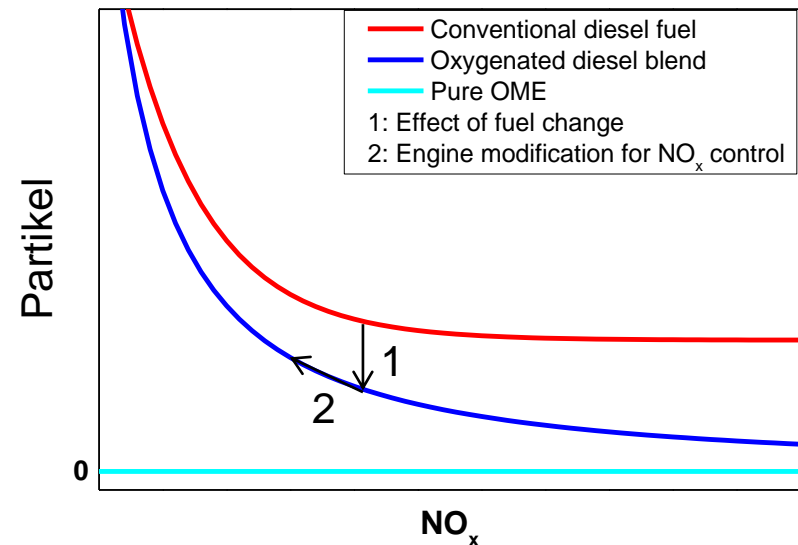
„Synthetic High Performance Fuel Components“

Elements

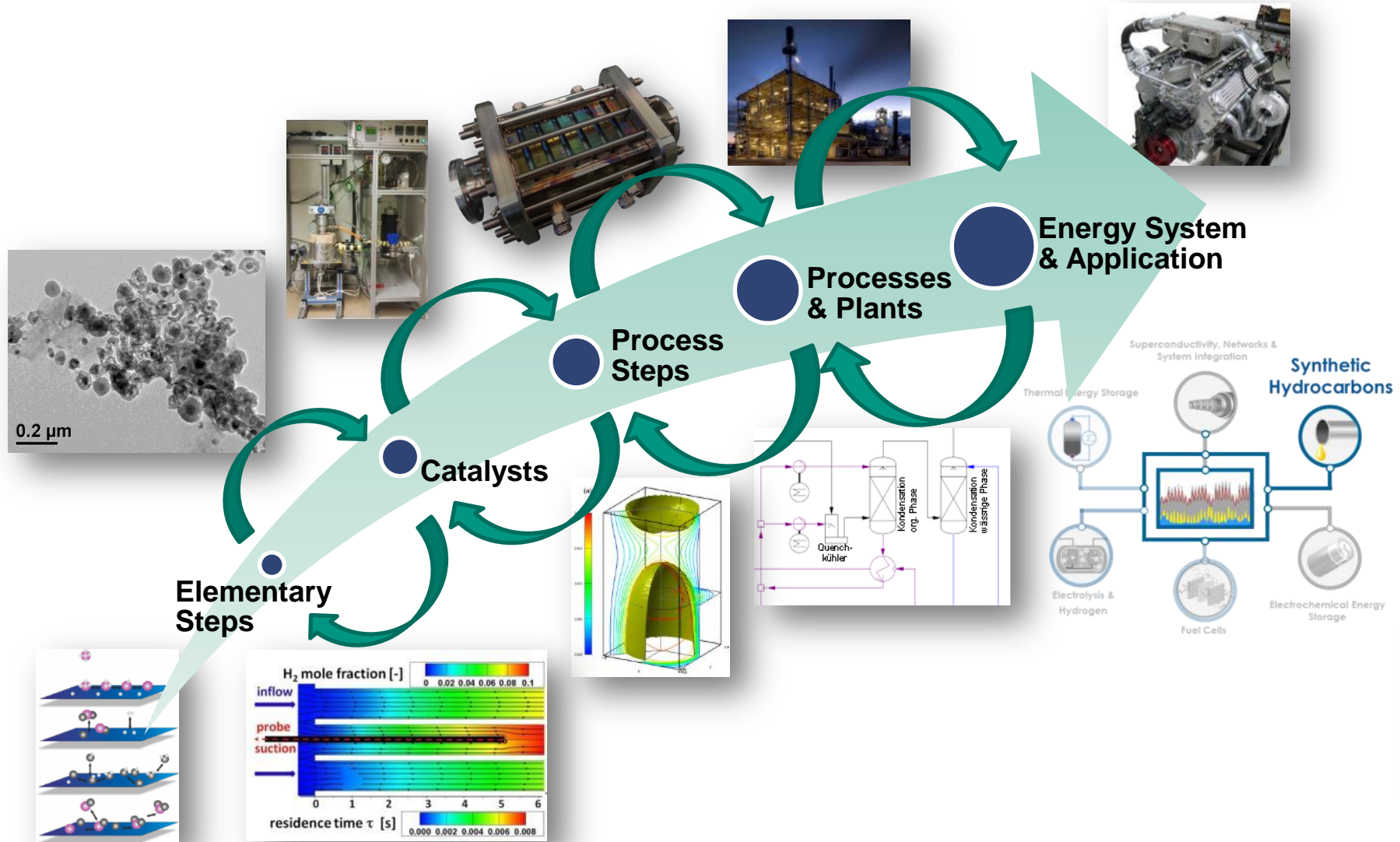
- Compatible to present fuels (drop-in)
- Reduced emissions
- Increased performance
- Reduced fuel consumption
- Reduced CO₂-footprint



Bsp.: Oxymethylenether (OME)



From Fundamentals to Applications and the Integration into the Energy System



Acknowledgements

- Sponsors and Funding Agencies
- Partners from Industry and Academia
- The teams from KIT
- The audience for your kind attention



GEFÖRDERT VOM



Gefördert durch:



aufgrund eines Beschlusses
des Deutschen Bundestages



Fachagentur Nachwachsende Rohstoffe e.V.

